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1.1 How to write a lab
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Lab 5 Chemistry 117

University of

Masachusetts Boston

Ser II out the worksheet

below, and use it as a

guide to decipher your

unknowns. Using the

column that will form a

precipitate when mixed.

Draw a dashed arrow

for gas formations and

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Fill a dotted arrow for
other types of reactions.
Connect Anion Formula
Reacting Ions
Compound Compound
Name Cation Formula
KOH HCl Vaa t LiBr
Summarize what ...

Solved: Lab 5
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University Of
Massachusetts Bos ...
CHEM 117 FINAL,

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Exam, Texas A&M
(TAMU) Chem Lab 117
Final Study Guide.

Calibration. Control
Experiment.

Reproducible. Density
of water decreases as
temperatu

Experimental
parameters used to
relate to fundamental
quantiti Shows whether
or not a result is due to

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the experimental facto

Multiple trials of several
measurements to verify
results. Graph on Page 5

...

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Laboratory Manual,
2019 Revision 117 3)
Reaction of Hydrogen

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with Oxygen: Text book
AHreaction per mole of
H₂O Formed: -286
K₃/mol Give a reference
for this value, making
sure you cite author,
year, text name and
edition, publisher, and
page number. Halg) t
Ogs HaOu AH H₀ =
-286K5lm Reaction of
Magnesium with
Oxygen: Summarize
Data from the first three

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Answers

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Answered: General
Chemistry I Laboratory
Manual, | bartleby
General Chemistry
Laboratory II: CHEM
117 Lab Manual builds
upon concepts and
techniques that were
presented in the first
semester of the General
Chemistry Laboratory
course as well as

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introducing the students
to labs that will help
them prepare for more
advance Chemistry
courses.

General Chemistry
Laboratory II: Chem
117 Lab Manual ...
CHEM 1211L Lab
Manual-Page 117
DATA AND
CALCULATIONS
(Show all data in an

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easy to follow format,
like a table. Attach
copies of your graphs)

Dilution Calculations

6-0-1 Data Reach

mass 0-120 (1) Write

Reaction A between

magnesium and

hydrochloric acid: (2)

Write Reaction B

between solid

magnesium oxide and

hydrochloric acid (3)

Write Reaction C for the

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formation of liquid
water AH3 286 ...

Experiment 5
Solved: CHEM 1211L
Lab Manual-Page 117
DATA AND
CALCULATIO ...

Pre-lab Questions: Read
the relevant textbook
sections and the entire
lab before answering pre-
lab questions. 1. If a
measured quantity is
written correctly, which

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Manual
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Experiment 5
digits are certain?

Which are uncertain? 2.

A new penny has a mass of 2.49 g and a volume of 0.349 cm³. Is the

penny pure copper? (density of copper = 8.96 g/cm³)

Show your work! 3.

Define ...

CHEM 100 Lab Manual

Harrisburg Area

Community College

You are required to

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Experiment 5

wear approved eye protection in the laboratory whenever you are doing any experiment or whenever any experiment is being done in the laboratory around you (eye protection must meet ANSI Z87.1 impact standards). Repeated failure to wear approved eye protection will result in dismissal from

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the experiment being
conducted. 2.

Experiment 5 GENERAL

CHEMISTRY 101

LABORATORY

MANUAL

Here is an example of a well-written abstract for the second lab in CHEM 117. The density of small polystyrene balls was determined from using the volume

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Manual in the
previous week's
experiment and the mass
of the ball. The
precisions of four
different strategies for
obtaining the mass of
the ball were measured
and compared.

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chapter questions and
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